Basic statistical tools for analytical laboratories

# Chapter 2 Basic statistical concept of measurement uncertainty

#### Understand the terms related to "Uncertainty"

• Consider expression like:

*"100.0 ml* <u>+</u> *0.15 ml with 95% confidence"* 

- The <u>+</u> 0.15 ml is the *uncertainty (U)* of the supposed 100.0 ml volume of the volumetric flask measured.
- This uncertainty (sometimes also known as **expanded uncertainty**) is actually estimated by having multiplied a **coverage factor** k with a **standard uncertainty** (u), expressed as standard deviation. i.e. U = k \* u
- The coverage factor *k* can be 2 or 3 for 95% or 99% confidence level, respectively
- **Relative standard uncertainty** = relative standard deviation or coefficient of variation CV,  $\frac{S}{X}$

#### **Important Note**

"Standard uncertainty is expressed as standard deviation"

#### Understand the terms related to "Uncertainty"

uncertainty range :  $y - U \dots y + U$  with 95% confidence interval **Error** = y - TV(y)mean y value y + U **y** – **U** | TV(y)(True Value)



# **Propagation Law of Standard Deviations**

- A test procedure, y involves many steps and each step can have a standard uncertainty, expressed as standard deviation, say  $x_i$ , then :  $y = f[x_1, x_2, x_3, ..., x_n]$
- The combined or total uncertainty of independent components is:

$$s(y)^{2} = \sum_{i=1}^{n} \left[\frac{\partial f}{\partial x_{i}}\right]^{2} s(x_{i})^{2}$$

•  $s(y)^2 = \sum \{ [\partial f/\partial x_1]^2 s(x_1)^2 + [\partial f/\partial x_2]^2 s(x_2)^2 + [\partial f/\partial x_3]^2 s(x_3)^2 + \dots \}$ 

# **Propagation Law of Standard Deviations**

 If the components of uncertainty or errors are NOT independent, there is an extra covariance factor to be considered:



# **Example of calculation**

 Let y (with uncertainty uy)has the following relationship with c, v and w which have standard uncertainty of u(c), u(v) and u(w), respectively:

$$y = \frac{c \times v \times 1000}{w}$$

• Differentiation of y with respect to c, v and w give:

$$\frac{dy}{dc} = \frac{v \times 1000}{w} \qquad \frac{dy}{dv} = \frac{c \times 1000}{w} \qquad \frac{dy}{dw} = -\frac{c \times v \times 1000}{w^2}$$

### **Example of calculation**

Given c = 0.45mg/L, v = 10ml, w = 1.5682g, and std uncertainty u(c) = 0.05mg/L, u(v) = 0.08ml, u(w) = 0.002g, then

$$y = \frac{c \times v \times 1000}{w} = \frac{0.45 \times 10 \times 1000}{1.5682} = 2,870 \,\mu g \,/\,L$$

• Individual sensitive coefficient is :

$$c_{c} = \frac{dy}{dc} = \frac{10 \times 1000}{1.5682} = 6376.74 \qquad c_{v} = \frac{dy}{dv} = \frac{0.45 \times 1000}{1.5682} = 286.95$$
$$c_{w} = \frac{dy}{dw} = -\frac{0.45 \times 10 \times 1000}{1.5682^{2}} = -1829.83$$

# **Example of calculation**

• Hence, the combined standard uncertainty of y, uy is calculated as follows:

• 
$$u_y^2 = \{c_c^2 u(c)^2 + c_v^2 u(v)^2 + c_w^2 u(w)^2\}$$
  
=  $\{6376.74^2 x 0.05^2 + 286.95^2 x 0.08^2 + (-1829.83)^2 x 0.002^2\} = 102197$ 

• 
$$u_y = 320 \mu g/L$$

# **Alternative approach**

- Some may find the above calculus calculation messy. However, it can be simplified as below:
- Let y has a relationship with A, B and C with a constant m as follows:  $y = m \frac{AB}{C}$
- By differentiation, we get:

$$u_{y}^{2} = \left(m\frac{B}{C}\right)^{2} u_{A}^{2} + \left(m\frac{A}{C}\right)^{2} u_{B}^{2} + \left(-m\frac{AB}{C^{2}}\right)^{2} u_{C}^{2}$$

# **Alternative approach**

- By dividing the above equation on both sides with  $\left(m\frac{AB}{C}\right)^2$
- we get:  $\left(\frac{u_y}{y}\right)^2 = \left(\frac{u_A}{A}\right)^2 + \left(\frac{u_B}{B}\right)^2 + \left(\frac{u_C}{C}\right)^2$
- where *u*'s are the standard uncertainties of *y*, *A*, *B* and *C*, respectively.
- Note that each of the above component is in fact their own coefficient of variation (CV).

### Propagation Law of Standard Deviation simplified

- Linear Combination of standard deviations
- y = K(a + b c) with std deviations  $s_a$ ,  $s_b$ ,  $s_c$  and K is a constant factor (or a fixed value such as 100),
- then,
- the combined standard deviation s<sub>v</sub> is
- $s_y = K * \sqrt{(s_a^2 + s_b^2 + s_c^2)}$
- Note : components *a*, *b* and *c* must be of same unit and involve in addition / subtraction

# Example of Linear Combination of standard deviations

- Combined uncertainty,
  U =
- $\sqrt{(0.0015)^2 + (0.0015)^2}$
- = 0.0021 gm

ertainty,			
		Weight, gm	Uncertainty, gm
0.0015/2	Wt of crucible + sample	45.3426	0.0015
	Wt of crucible	40.2325	0.0015
	Wt of sample	5.1101	?
			14

#### Propagation Law of Random Errors

- Multiplicative Combination
- y = K(a \* b) / c with std deviations  $s_a$ ,  $s_b$ ,  $s_c$ and K is a constant factor,
- Then,
- the combined standard deviation  $s_v$  is
- $s_y / y = \sqrt{[(s_a / a)^2 + (s_b / b)^2 + (s_c / c)^2]}$
- Why is there no *K* in the above equation?

# **Worked Example**

• 
$$y = (A * B) / C$$

	Value, V	Uncertainty U
A	22.5 mg/L	0.2 mg/L
В	250 ml	1.2 ml
C	100 gm	0.8 gm
У	56.25 mg/kg	?

- $U_y / y = \sqrt{(u_a/A)^2 + (u_b/B)^2 + (u_c/C)^2}$
- Answer :  $U_y = 0.725 \text{ mg/kg}$